

Fe-Ga-Gd (Iron-Gallium-Gadolinium)

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Recently, [2009Liu] determined an isothermal section at 500 °C for this ternary system, which depicts two ternary compounds along the isoconcentrate line of 7.7 at.% Gd.

Binary Systems

The Fe-Ga phase diagram [2004Oka, Massalski2] has the following intermediate phases: α' (B_2 , $CsCl$ -type cubic), α'' ($D_0{}_3$, BiF_3 -type cubic), αFe_3Ga ($L1_2$, $AuCu_3$ -type cubic), βFe_3Ga ($D0_{19}$, Ni_3Sn -type hexagonal), αFe_6Ga_5 (Fe_6Ge_5 -type monoclinic), βFe_6Ga_5 ($D8_{10}$, Al_8Cr_5 -type rhombohedral), Fe_3Ga_4 (monoclinic) and $FeGa_3$ ($CoGa_3$ -type tetragonal). The Fe-Gd phase diagram [1998Zha] depicts the following compounds: $\alpha Fe_{17}Gd_2$ (Th_2Zn_{17} -type rhombohedral), $\beta Fe_{17}Gd_2$ (Th_2Ni_{17} -type hexagonal), $Fe_{23}Gd_6$ ($D8_a$, $Mn_{23}Th_6$ -type cubic), Fe_3Gd (Ni_3Pu -type rhombohedral) and Fe_2Gd ($C15$, $MgCu_2$ -type cubic). The temperature range of stability of $Fe_{23}Gd_6$ is not firmly established [1998Zha]. [2009Liu] did not find this

compound at 500 °C. The Ga-Gd phase diagram [Massalski2] has the following intermediate phases: $GdGa_6$ ($PuGa_6$ -type tetragonal), $GdGa_2$ (22-33.3 at.% Gd; $C32$, AlB_2 -type hexagonal), $GdGa$ (B_f , CrB -type orthorhombic), Gd_3Ga_2 (Gd_3Ga_2 -type tetragonal) and Gd_5Ga_3 ($D8_l$, Cr_5B_3 -type tetragonal).

Ternary Compounds

Two ternary compounds are stable in this system at 500 °C [2009Liu]: $GdFe_{5.3}Ga_{6.7}$ ($ScFe_6Ga_6$ -type orthorhombic, space group $Imm\bar{m}$, lattice parameters: $a = 0.85676$ nm, $b = 0.86960$ nm and $c = 0.50782$ nm, denoted as τ_1 here and as δ by [2009Liu]) and $GdFe_5Ga_7$ ($ThMn_{12}$ -type tetragonal, $a = 0.8651$ nm and $c = 0.50934$ nm, denoted τ_2 here and ε by [2009Liu]). Both these compounds are present at a constant Gd content of 7.7 at.%. $GdFe_5Ga_7$ shows a homogeneity range of 53.8-59.2 at.% [2009Liu], which corresponds to a range of $GdFe_5Ga_7$ - $GdFe_{4.3}Ga_{7.7}$.

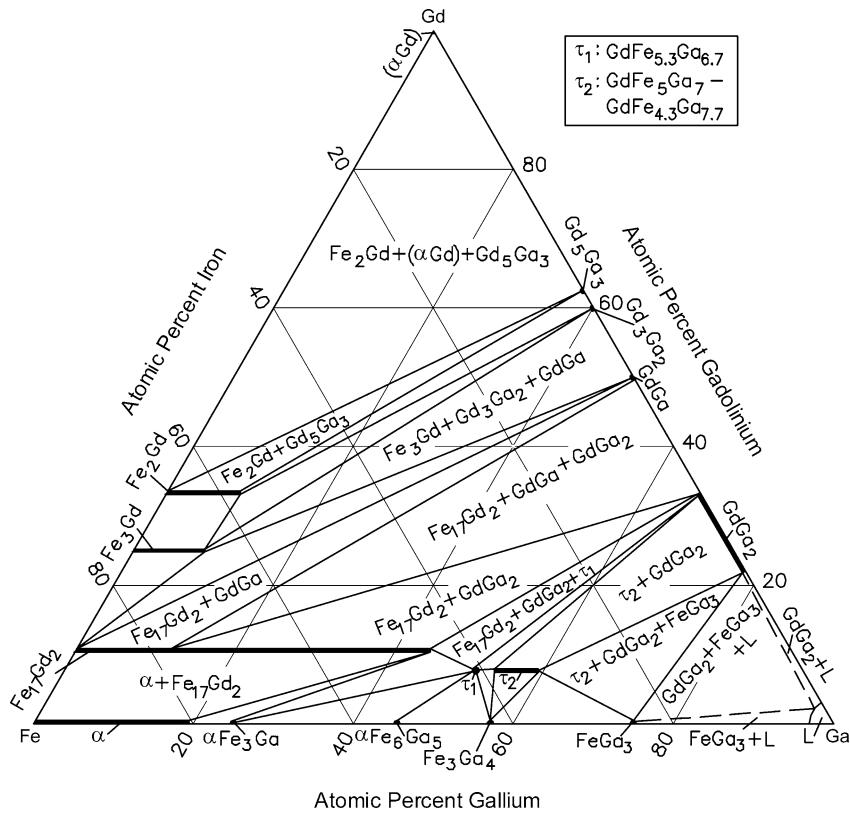


Fig. 1 Fe-Ga-Gd isothermal section at 500 °C [2009Liu]. Narrow two-phase regions are omitted

Section II: Phase Diagram Evaluations

Isothermal Section

With starting metals of 99.99+% purity, [2009Liu] arc-melted about 130 alloys under Ar atm. The samples were given a final anneal at 500 °C for 3 days and quenched in liquid nitrogen. The phase equilibria were studied with x-ray powder diffraction. The isothermal section constructed by [2009Liu] at 500 °C is shown in Fig. 1. The ternary compounds τ_1 and τ_2 are present. The binary compounds Fe₂Gd, Fe₃Gd and α Fe₁₇Gd₂ dissolve up to 9.2, 9 and 44.3 at.% Ga respectively. The other binary compounds do not show any ternary solubility. The compound GdGa₆ is not stable at this temperature. Fe₂₃Gd₆ was not detected at 500 °C [2009Liu]. No homogeneity ranges were reported

by [2009Liu] for α Fe₃Ga, α Fe₆Ga₅ and Fe₃Ga₄ binary compounds.

References

- 1998Zha:** W. Zhang, C. Li, X. Su, and K. Han, An Updated Evaluation of the Fe-Gd (Iron-Gadolinium) System, *J. Phase Equilib.*, 1998, **19**(1), p 56-63
- 2004Oka:** H. Okamoto, Fe-Ga (Iron-Gallium), *J. Phase Equilb. Diffus.*, 2004, **25**(1), p 100
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